

## **ENTHALPY OF VAPORIZATION OF WATER**

The enthalpy (heat) of vaporization of water is tabulated as a function of temperature on the IPTS-68 scale.

### **REFERENCE**

Marsh, K. N., Ed., *Recommended Reference Materials for the Realization of Physicochemical Properties*, Blackwell, Oxford, 1987.

$t$ °C	$\Delta_{\text{vap}}H$ kJ/mol	$t$ °C	$\Delta_{\text{vap}}H$ kJ/mol
0	45.054	200	34.962
25	43.990	220	33.468
40	43.350	240	31.809
60	42.482	260	29.930
80	41.585	280	27.795
100	40.657	300	25.300
120	39.684	320	22.297
140	38.643	340	18.502
160	37.518	360	12.966
180	36.304	374	2.066